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# RADIOACTIVITY SURVEY DATA in Japan

NUMBER 77 June 1986

National Institute of Radiological Sciences Chiba, Japan

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# Number 77

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### Contents

			Page	
Envi	ronmer	Ital and Dietary Materials	1	
	1	Japan Chemical Analysis Center)		
1.	Collect	tion and Pretreatment of Samples	1	
2.		ration of Samples for Analysis		
3.	Separa	ation of Strontium-90 and Cesium-137	3	
4.		nination of Stable Strontium, Calcium and Potassium		
5.		nq		
6.		s		
-	(1)	Strontium-90 and Cesium-137 in Total Diet	_	
	` '	Strontium-90 and Cesium-137 in Rice		
	(-, -	(producing districts)		
	-2	Strontium-90 and Cesium-137 in Rice	8	
	_	(consuming districts)	• •	
	(3)-1	Strontium-90 and Cesium-137 in Milk	g	i
	(0)	(producing districts for domestic program)	•	
	-2	Strontium-90 and Cesium-137 in Milk	10	1
		(producing districts for WHO program)		
	-3	Strontium-90 and Cesium-137 in Milk	11	
	-3	(consuming districts)		
		Strontium-90 and Cesium-137 in Milk	12	
	-4		12	
	/4\ 1	(powder milk)	12	,
	(4)-1	Strontium-90 and Cesium-137 in Vegetables	13	,
	•	(producing districts)	4.4	1
	-2	Strontium-90 and Cesium-137 in Vegetables	14	,
	<b></b> .	(consuming districts)		
	(5)	Strontium-90 and Cesium-137 in Tea (Japanese Tea)		
	(6)	Strontium-90 and Cesium-137 in Sea Fish		
	(7)	Strontium-90 and Cesium-137 in Shellfish		
	(8)	Strontium-90 and Cesium-137 in Seaweeds	19	ĺ

### **Environmental and Dietary Materials\***

(Japan Chemical Analysis Center)

### 1. Collection and pretreatment of samples

### (1) Rain and dry fallout

Rain and dry fallout was collected monthly on a sampling tray, approximately  $5000~\rm{cm}^2$  in area, which was filled with water to a depth of 1 cm at the beginning of every month.

Strontium and cesium carrier solutions were added after the sample was filtered. The tray was washed with  $5\ell$  of distilled water and the washing was combined to the filtrate.

The sample was passed through a cation exchange column (500 m $\ell$  of Dowex 50W X8, 50 $\sim$ 100 mesh, Na form) at a rate flow of 80 m $\ell$ /min.

### (2) Airborne dust

Airborne dust was collected by an electrostatic precipitator or a filter air sampler for every three months at a rate of more than 3000 m<sup>3</sup> per month. The sampling was done 1 to 1.5 meters above the ground.

### (3) Service water and freshwater

Service water, 100  $\ell$  each, was collected at the intake of the water-treatment plant and at the tap after water was left running for five minutes. Strontium and cesium carriers were added to the filtered water sample. The subsequent process was the same as that described in the section (1). Freshwater was treated in the same way as the service water.

### (4) Soil

Soil was collected from the location in the spacious and flat area without past surface disturbance caused by duststorms, inflow and outflow due to precipitation, etc.. Any places located under trees in a forest, in a stony area or inside of river banks were avoided. Soil was taken from two layers of different depths,  $0\sim5$  cm and  $5\sim20$  cm. The soil lumps were crushed by hands and dried in a drying oven regulated  $105^{\circ}$ C. The soil was then passed through a 2 mm sieve to remove plant roots and pebbles.

### (5) Sea water

Sea water was collected at the fixed stations

where the effect of terrestrial fresh water from rivers was expected to be negligibly small. A special consideration was also given to weather conditions. The sampling was carried out when there was no rainfall for the last few days. To prevent contamination, water samples were collected at the bow of a sampling boat just before she stood still by scooping surface water using a polyethylene bucket. Immediately after the collection, the samples were acidified to a pH lower than 3 by adding concentrated hydrochloric acid in a ratio of 1  $m\ell$  to  $1\ell$  of sea water, and then stored in  $20-\ell$ polyethylene containers. The sampling equipments as well as containers were thoroughly rinsed with dilute hydrochloric acid and then with distilled water before use. Two hundred milliliters of sea water was also collected at the same stations for the determination of chlorinity.

### (6) Sea sediments

Sediment was collected in the same area as that for the sea water sample, taking the following criteria into account:

- a. The depth of water exceeds 1 m at low tide.
- No significant sedimental movement is observed in the vicinity of concern.
- c. Mud, silt and fine sand are preferable.

A conventional sediment sampling device was used for collecting the top few centimeters of surface sediment. Approximately 4 kg of the sample in wet weight was spread on a stenless steel dish after removed of the pebbles, shells and other foreign materials, and dried in a drying oven regulated at 105°C.

### (7) Total diet

A full one day ordinary diet including three meals, water, tea and other in-between snacks for five persons was collected as a sample of "total diet". The sample in a large stainless steel pan was carbonized carefully by direct application of gas flame, and was transfered to a porcelain dish and then ashed at 450  $^{\circ}$ C in an electric muffle furnace.

### (8) Rice

Polished rice was collected in producing districts at the harvest and in consuming areas when new crops were first put on sale. The sample was carbonized and ashed in a porcelain dish.

<sup>\*</sup> Samples were sent to the Center from 32 contracted prefectures.

### (9) Milk

Raw milk was collected in producing districts and commercial milk was purchased in consuming districts. Milk in a stainless steel pan or a porcelain dish was evaporated to dryness followed by carbonization and ashing.

### (10) Vegetables

Spinach and Japanese radish were selected as the representatives for leaf vegetables and for nonstarch roots, respectively. After removing soil, the edible part of vegetable sample was dried and carbonized in a stainless steel pan or a porcelain dish.

### (11) Tea

Five hundred grams of manufactured green tea was collected, carbonized and ashed in a stainless steel pan or a porcelain dish.

### (12) Fish, shellfish and seaweeds

### a. Sea fish and freshwater fish

Fish was rinsed with water and blotted with a filter paper. Only the edible part was used in case of larger sized fish, and the whole part was used in case of smaller ones. Each sample was weighed and placed in a stainless steel pan or a porcelain dish. After carbonized, the sample was ashed in an electric muffle furnace.

### b. Shellfish

Approximately 4 kg of shellfish including the shells was collected or purchased. After removing the shells, it was treated in the same way as that for the sea fish.

### c. Seaweeds

Edible seaweeds were collected and rinsed with water to remove sand and other adhering matters on the surface. These were removed of excess water, weighed dried and ashed.

Table 1 shows detailes of sample collection.

Table 1 Details of sample collection

Sample	Frequency of sampling	Quantity of sample
=Environmental materials=		
(1) Rain and dry fallout		•
1 For domestic program	monthly	
2 For WHO program	monthly	
(2) Airborne dust	quarterly	$>$ 3000 m $^3$ /month
(3) Service water and freshwater		
1 Service water (source water)	semiyearly	100ℓ
2 Service water (tap water)	semiyearly	100ℓ
3 Freshwater	yearly (fishing season)	100ℓ
(4) Soil		
1 0~5 cm	yearly	4 kg
2 5~20cm	yearly	4 kg
(5) Sea water	yearly	40ℓ
(6) Sea sediments	yearly	4 kg
=Dietary materials=		
(7) Total diet	semiyearly	daily amount for 5 person
(8) Rice		
1 Producing districts		
-	yearly (harvesting season)	
		5 kg (polished rice)
2 Consuming districts	yearly (harvesting season)	5 kg (polished rice)
(9) Milk		
1 Producing districts for	guarterly (February, May, August and	3 ℓ
WHO program	November)	
2 Producing districts for domestic program	semiyearly (February and August)	3ℓ

	Sample	Frequency of sampling	Quantity of sample
3	Consuming districts	semiyearly (February and August)	3ℓ
4	Powdered milk	semiyearly (April and October)	2~3 kg
(10)	Vegetables		
1	Producing districts	yearly (hervesting season)	4 kg
2	Consuming districts	yearly (harvesting season)	4 kg
(11)	Геа	yearly (the first harvesting season)	500 g (manufactured tea)
(12)	Fish, shellfish, and seaweeds		
1	Sea fish	yearly (fishing season)	4 kg
2	Freshwater fish	yearly (fishing season)	4 kg
3	Shellfish	yearly (fishing season)	4 kg
4	Seaweeds	yearly (fishing season)	2~3 kg

### 2. Preparation of samples for analysis

### (1) Rain, service water and freshwater

Strontium and cesium were eluted with hydrochloric acid from the cation exchange column. The residue of rain sample on the filter paper was ashed in an electric muffle furnace and the ash was dissolved in hydrochloric acid. The insoluble part was filtered and washed. The filtrate and the washings were combined to the previous eluate and used for radiochemical analysis.

### (2) Soil and Sea sediment

Dried soil was crushed to smaller ones than 0.25 mm in size by a crusher. The sieved sample was ashed in an electric muffle furnace regulated at  $450^{\circ}$ C. The sample was then heated with hydrochloric acid, strontium and cecium carrier solutions and the mixture was heated. The insoluble constituent was filtered off and washed with water.

The dried sample was crushed to smaller ones than 0.25 mm by a crushing machine. The further preparation of the sample was the same as that described in the section 2-(2).

### (3) Rice

The ashed sample was pulverized with a porcelain mortar and passed through a 0.35 mm sieve. The sieved sample to which both strontium and cesium carriers were added, was digested with nitric acid by heating. After the sample was heated again with nitric acid to dryness, strontium and cesium were extracted with hydrochloric acid and water. The insoluble constituent was filtered and washed. The filtrate and washings were combined for subsequent radiochemical analysis.

### (4) Airborne dust, diet, milk, vegetables, fish and shellfish, seaweeds, tea, and others

These ashed samples were treated with the same procedure as that described in the section 2-(4).

### 3. Separation of strontiunm-90 and cesium-137

### (1) Strontium-90

Sample solutions, prepared as in the foregoing sections 2-(1) through 2-(5), were neutralized with sodium hydroxide. After sodium carbonate was added, the precipitate of strontium and calcium carbonates was separated. The supernatant solution was retained for cesium-137 determination. The carbonates were dissolved in hydrochloric acid and strontium and calcium were precipitated as oxalates. The precipitate was dissolved in nitric acid and strontium was separated from calcium by successive fuming nitric acid separation. Iron scavenge was made after addition of ferric iron carrier followed by barium chromate separation after addition of barium carrier to remove radium, its daughters and lead. Strontium was recovered as carbonate, and the precipitate was dried and weighed to determine strontium recovery. The strontium carbonate was dissolved in hydrochloric acid and iron carrier was added. The solution was allowed to stand for two weeks for strontium-90 and yttrium-90 to attain equilibrium. Yttrium-90 was coprecipitated with ferric hydroxide and the precipitate was filtered off, washed and counted.

### (2) Cesium-137

The supernatant separated from the strontium fraction was acidified with hydrochloric acid. While stirring, cesium was adsorbed on the ammonium molyb-

dophosphate added.

After filtered off and washed with hydrochlotric acid the precipitate was dissolved in 2.5 N sodium hydroxide solution. The solution was adjusted to pH 8.2 with hydrochloric acid and allowed to cool. Resultant molybdenum hydroxide which separated out in the solution, was filtered off and washed with water. EDTA was added to the filtrate and washings. Cesium and rubidium were adsorbed on a cation exchange column and cesium was separated from rubidium by eluting with hydrochloric acid.

The eluate was evaporated to dryness and was dissolved. The solution was filtered. Chloroplatinic acid was added to precipitate cesium. The precipitate was filtered onto a tared paper using a demountable filter and washed with water and then ethanol. After drying, the chemical yield of cesium was determined by weighing the precipitate. Cesium-137 radioactivity was measured for this precipitate.

4. Determination of stable strontium, calcium and potassium

A weighed amount of soil or sea sediment was

heated in a electric muffle furnace at 450°C and then treated with hydrochloric acid for extraction. A weighed aliquot of ashed samples of total diet, vegetables, milk, fish, shellfish or seaweeds was digested with hydrofloric acid and nitric acid. The extract was made up to an appropriate volume with dilute hydrochloric acid. The sample solution was analyzed for calcium by titration with standard potassium permanganate solution after separating calcium as oxalate. Atomic absorption spectroscopy was applied when appropriate. Stable strontium and potassium were determined by atomic absorption and flame emission spectrometry, respectively.

### 5. Counting

After the radiochemical separation, the mounted precipitates were counted for activity using low background beta counters normally for 60 to 90 min. Net sample counting rates were corrected for counter efficiency, recovery, self-absorption and decay to obtain the content of strontium-90 and cesium-137 per sample aliquot. From the results, concentrations of these nuclides in the original samples were calculated.

### 6. Results

# (1) Strontium-90 and Cesium-137 in Total Diet (from Nov. 1985 to Jul. 1986)

Table (1): Strontium-90 and Cesium-137 in Total Diet

Longhian	Ash	Ca	K	90	'Sr	137	Cs
Location	(g/p·d)	(mg/p·d)	(mg/p·d)	(pCi/p·d)	S.U.	(pCi/p·d)	C.U.
November, 1985							
Nishikawa-machi, NIIGATA	18.9	598	2580	$4.3 \pm 0.43$	$7.2 \pm 0.72$	$1.6 \pm 0.25$	$0.6 \pm 0.10$
Kanazawa, ISHIKAWA	18.8	609	2650	$4.1 \pm 0.44$	$6.7 \pm 0.72$	$2.0 \pm 0.26$	$0.8 \pm 0.10$
Fukui, FUKUI	14.1	675	1600	$2.1 \pm 0.27$	$3.1 \pm 0.40$	$0.6 \pm 0.16$	$0.4 \pm 0.10$
Neyagawa, OSAKA	16.5	665	2150	$2.2 \pm 0.33$	$3.3 \pm 0.49$	$2.5 \pm 0.28$	$1.2 \pm 0.13$
Matsue, SHIMANE	25.2	1400	3230	$4.0 \pm 0.50$	$2.9 \pm 0.36$	$4.1 \pm 0.40$	$1.3 \pm 0.12$
Nagasaki, NAGASAKI	13.6	435	1830	$1.3 \pm 0.26$	$3.0 \pm 0.60$	1.6 ± 0.19	$0.8 \pm 0.10$
December, 1985							
Aomori, AOMORI	18.3	437	2100	$3.2 \pm 0.38$	$7.3 \pm 0.86$	$1.1 \pm 0.22$	$0.5 \pm 0.10$
Shinjuku, TOKYO	14.6	375	1810	$1.9 \pm 0.26$	$5.0 \pm 0.71$	$1.6 \pm 0.20$	$0.9 \pm 0.11$
Nagano, NAGANO	18.6	792	2530	$3.0 \pm 0.38$	$3.8 \pm 0.48$	$1.3 \pm 0.24$	$0.5 \pm 0.09$
Kyoto, KYOTO	17.4	812	2160	$2.2 \pm 0.36$	$2.8 \pm 0.45$	$1.5 \pm 0.27$	$0.7 \pm 0.13$
January, 1986							
Fukushima, FUKUSHIMA	18.4	488	2180	$2.9 \pm 0.36$	$5.9 \pm 0.73$	$2.8 \pm 0.28$	$1.3 \pm 0.13$
Hiroshima, HIROSHIMA	11.9	857	1340	$1.3 \pm 0.21$	$1.5 \pm 0.24$	$2.0 \pm 0.18$	$1.5 \pm 0.14$
Yamaguchi, YAMAGUCHI	20.2	608	2570	$2.5 \pm 0.41$	$4.1 \pm 0.68$	$2.5 \pm 0.29$	$1.0 \pm 0.11$
Naha, OKINAWA	17.1	681	2150	$1.2 \pm 0.30$	$1.8 \pm 0.45$	$1.4 \pm 0.21$	$0.7 \pm 0.10$
May, 1986							
Kanazawa, ISHIKAWA	16.9	608	2050	$2.4 \pm 0.35$	$3.9 \pm 0.57$	$6.0 \pm 0.37$	$2.9 \pm 0.18$
Wakayama, WAKAYAMA	8.90	296	911	$0.8 \pm 0.16$	$2.8 \pm 0.56$	$1.6 \pm 0.15$	$1.8 \pm 0.16$
June, 1986							
Sapporo, HOKKAIDO	18.4	1030	2430	$5.0 \pm 0.40$	$4.9 \pm 0.39$	$8.5 \pm 0.43$	$3.5 \pm 0.18$
Aomori, AOMORI	20.3	1280	2210	$3.4 \pm 0.41$	$2.7 \pm 0.32$	$5.4 \pm 0.38$	$2.4 \pm 0.17$
Yamagata, YAMAGATA	15.8	542	1680	$3.4 \pm 0.34$	$6.3 \pm 0.63$	$3.1 \pm 0.26$	$1.9 \pm 0.15$
Mito, IBARAGI	17.5	562	2060	$3.2 \pm 0.36$	$5.7 \pm 0.64$	$3.7 \pm 0.29$	$1.8 \pm 0.14$
Shinjuku, TOKYO	13.0	404	1980	$1.3 \pm 0.25$	$3.2 \pm 0.63$	$4.0 \pm 0.26$	$2.0 \pm 0.13$
Nishikawa-machi, NIIGATA	17.1	662	2420	$2.9 \pm 0.38$	$4.3 \pm 0.57$	$7.6 \pm 0.42$	$3.1 \pm 0.17$
Fukui, FUKUI	13.8	699	1500	$2.4 \pm 0.27$	$3.4 \pm 0.39$	$3.0 \pm 0.24$	$2.0 \pm 0.16$
Nagano, NAGANO	16.4	833	2130	$2.4 \pm 0.30$	$2.9 \pm 0.36$	$3.2 \pm 0.26$	$1.5 \pm 0.12$
Shizuoka, SHIZUOKA	15.9	742	2530	$2.0 \pm 0.29$	$2.7 \pm 0.38$	$4.7 \pm 0.31$	$1.9 \pm 0.12$
Nagoya, AICHI	22.1	1170	2290	$2.6 \pm 0.39$	$2.2 \pm 0.33$	$6.9 \pm 0.43$	$3.0 \pm 0.19$
Kyoto, KYOTO	16.4	538	2190	2.1 ± 0.35	3.9 ± 0.65	$2.3 \pm 0.27$	$1.0 \pm 0.12$

Location	Ash	Ash Ca K		9 0	'Sr	137 <sub>Cs</sub>		
Location	(g/p·d)	(mg/p·d)	(mg/p·d)	(pCi/p·d)	S.U.	(pCi/p·d)	C.U.	
Kakogawa, HYOGO	12.1	732	1460	1.6 ± 0.22	2.2 ± 0.31	2.5 ± 0.21	$1.7 \pm 0.14$	
Fukube-mura, TOTTORI	15.4	587	1730	$2.4 \pm 0.30$	$4.1 \pm 0.51$	$5.8 \pm 0.34$	$3.4 \pm 0.20$	
Okayama, OKAYAMA	15.6	477	1810	$1.9 \pm 0.30$	$4.0 \pm 0.63$	$6.0 \pm 0.35$	$3.3 \pm 0.19$	
Yamaguchi, YAMAGUCHI	13.5	475	1750	$1.7 \pm 0.24$	$3.6 \pm 0.51$	$4.5 \pm 0.28$	$2.6 \pm 0.16$	
Matsuyama, EHIME	13.2	496	1470	$1.5 \pm 0.26$	$3.0 \pm 0.52$	$2.4 \pm 0.23$	$1.6 \pm 0.15$	
Kochi, KOCHI	13.9	626	2060	$3.1 \pm 0.32$	$4.9 \pm 0.51$	15 ± 0.5	$7.1 \pm 0.24$	
Dazaifu, FUKUOKA	11.9	465	2020	$1.7 \pm 0.23$	$3.6 \pm 0.49$	$4.9 \pm 0.27$	$2.4 \pm 0.13$	
Saga, SAGA	14.1	674	1870	$1.8 \pm 0.27$	$2.7 \pm 0.40$	$2.9 \pm 0.25$	$1.6 \pm 0.13$	
Ookuchi, KAGOSHIMA	11.0	209	1310	$1.3 \pm 0.21$	$6.1 \pm 0.98$	$4.5 \pm 0.25$	$3.4 \pm 0.19$	
JUly, 1986								
Akita, AKITA	16.3	549	1950	$2.0 \pm 0.31$	$3.6 \pm 0.56$	$9.7 \pm 0.43$	$5.0 \pm 0.22$	
Hiratsuka, KANAGAWA	15.3	532	2420	$1.5 \pm 0.28$	$2.9 \pm 0.52$	$10 \pm 0.4$	$4.2 \pm 0.18$	

# (2)-1 Strontium-90 and Cesium-137 in Rice(producing districts) (from Nov. 1985 to Dec. 1985)

Table (2)-1: Strontium-90 and Cesium-137 in Rice

	Component			•°Sr		137Cs	
Location	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	S.U.	pci/Kg	c.u.
November, 1985 Ishikari-machi, HOKKAIDO	0.696	0.086	1.21	0.3 ± 0.19	4 ± 2.3	0.7 ± 0.14	0.6 ± 0.11
December, 1985 Fukushima, FUKUSHIMA Akashi, HYOGO Chikushino, FUKUOKA	0.688 0.540 0.574	0.072 0.062 0.059	0.997 0.783 0.883	$\begin{array}{ccc} 0.2 & \pm & 0.19 \\ 0.0 & \pm & 0.15 \\ 0.2 & \pm & 0.15 \end{array}$	$3 \pm 2.6$ $0 \pm 2.4$ $4 \pm 2.6$	$\begin{array}{ccc} 2.1 & \pm 0.22 \\ 0.2 & \pm 0.11 \\ 0.4 & \pm 0.13 \end{array}$	$\begin{array}{cccc} 2.1 & \pm 0.22 \\ 0.2 & \pm 0.14 \\ 0.5 & \pm 0.14 \end{array}$

# (2)-2 Strontium-90 and Cesium-137 in Rice(consuming districts) (from Nov. 1985 to Jan. 1986)

Table (2)-2: Strontium-90 and Cesium-137 in Rice

T	Component			90S1	•	<sup>137</sup> Cs		
Location	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	s.u.	pci/Kg	C.U.	
November, 1985								
Sapporo, HOKKAIDO	0.599	0.062	1.16	$0.4 \pm 0.17$	$6 \pm 2.7$	$0.7 \pm 0.13$	$0.6 \pm 0.11$	
Fukui, FUKUI	0.556	0.063	0.911	$0.3 \pm 0.16$	$5 \pm 2.5$	$0.6 \pm 0.13$	$0.6 \pm 0.15$	
December, 1985								
Kobe, HYOGO	0.566	0.063	0.905	$0.2 \pm 0.17$	$3 \pm 2.7$	$1.2 \pm 0.15$	$1.3 \pm 0.16$	
Matsue, SHIMANE	0.566	0.059	0.735	$0.3 \pm 0.18$	$5 \pm 3.1$	$6.7 \pm 0.30$	$9.0 \pm 0.41$	
Kasuga, FUKUOKA	0.627	0.064	0.858	$0.5 \pm 0.18$	$7 \pm 2.9$	$0.5 \pm 0.15$	$0.5 \pm 0.17$	
Yonagusuku-mura, OKINAWA	0.533	0.053	0.842	$0.2 \pm 0.15$	4 ± 2.8	$0.2 \pm 0.11$	$0.2 \pm 0.13$	
January, 1986								
Hirosaki, AOMORI	0.533	0.059	0.724	$0.2 \pm 0.14$	$3 \pm 2.4$	$0.3 \pm 0.11$	$0.5 \pm 0.15$	
Nagasaki, NAGASAKI	0.633	0.068	0.810	$0.1 \pm 0.17$	$1 \pm 2.5$	$1.4 \pm 0.18$	$1.8 \pm 0.22$	

# (3)-1 Strontium-90 and Cesium-137 in Milk(producing districts for domestic program) (from Feb. 1986 to Jun. 1986)

Table (3)-1: Strontium-90 and Cesium-137 in Milk

	Component			90	Sr	137Cs	
Location	Ash(g/1)	Ca(g/1)	K(g/1)	pCi/l	S.U.	pci/l	c.u.
February, 1986					,		
Aomori, AOMORI	7.60	1.26	1.57	$2.4 \pm 0.26$	$1.9 \pm 0.21$	$2.8 \pm 0.24$	$1.8 \pm 0.15$
Mito, IBARAGI	7.67	1.22	1.77	$0.5 \pm 0.20$	$0.4 \pm 0.17$	$1.0 \pm 0.18$	$0.6 \pm 0.10$
Oshimizu-machi, ISHIKAWA	7.12	1.13	1.63	$1.3 \pm 0.33$	$1.2 \pm 0.29$	$2.2 \pm 0.31$	$1.4 \pm 0.19$
Mihara-machi, HYOGO	7.13	1.09	1.60	$0.7 \pm 0.19$	$0.6 \pm 0.18$	$0.4 \pm 0.13$	$0.2 \pm 0.08$
Matsuyama, EHIME	8.18	1.31	1.66	$1.0 \pm 0.24$	$0.8 \pm 0.18$	$0.7 \pm 0.15$	$0.4 \pm 0.09$
June, 1986							
Yamato-machi, SAGA	7.22	1.12	1.55	$1.0 \pm 0.22$	$0.9 \pm 0.19$	$7.9 \pm 0.37$	$5.1 \pm 0.24$

# (3)-2 Strontium-90 and Cesium-137 in Milk(producing districts for WHO program) (from Dec. 1985 to Jun. 1986)

Table (3)-2: Strontium-90 and Cesium-137 in Milk

	Component			90	Sr	133	Cs Cs
Location -	Ash(g/1)	Ca(g/1)	K(g/l)	pCi/l	s.u.	pci/l	c.u.
December, 1985							
Yakumo-mura, SHIMANE	8.07	1.36	1.34	$1.6 \pm 0.27$	$1.1 \pm 0.20$	$3.8 \pm 0.30$	$2.8 \pm 0.22$
January, 1986							
Nose-machi, OSAKA	7.56	1.17	1.56	$0.5 \pm 0.19$	$0.4 \pm 0.16$	$0.5 \pm 0.15$	$0.3 \pm 0.10$
Takamiya-machi, HIROSHIMA	6.86	1.06	1.46	$1.0 \pm 0.20$	$0.9 \pm 0.19$	$0.8 \pm 0.17$	$0.5 \pm 0.11$
February, 1986							
Sapporo, HOKKAIDO	7.51	1.32	1.61	$2.0 \pm 0.26$	$1.5 \pm 0.20$	$2.5 \pm 0.23$	$1.6 \pm 0.14$
Hachijo-Island, TOKYO	6.88	1.12	1.56	$5.2 \pm 0.35$	$4.6 \pm 0.31$	$16 \pm 0.5$	$10 \pm 0.3$
Nishikawa-machi, NIIGATA	8.10	1.27	1.71	$0.9 \pm 0.24$	$0.7 \pm 0.19$	$0.6 \pm 0.18$	$0.3 \pm 0.11$
Katsuyama, FUKUI	7.73	1.23	1.64	$1.5 \pm 0.29$	$1.2 \pm 0.23$	$2.1 \pm 0.25$	$1.3 \pm 0.15$
Kochi, KOCHI	7.39	1.09	1.72	$1.5 \pm 0.27$	$1.3 \pm 0.25$	$0.9 \pm 0.18$	$0.5 \pm 0.11$
Fukuma-machi, FUKUOKA	7.66	1.21	1.48	$0.3 \pm 0.22$	$0.3 \pm 0.19$	$1.3 \pm 0.21$	$0.9 \pm 0.14$
Kajiki-machi, KAGOSHIMA	7.58	1.16	1.59	$0.8 \pm 0.21$	$0.6 \pm 0.18$	$1.6 \pm 0.22$	$1.0 \pm 0.14$
May, 1986							
Sapporo, HOKKAIDO	7.72	1.29	1.60	$2.4 \pm 0.27$	$1.9 \pm 0.21$	$37 \pm 0.8$	$23 \pm 0.5$
Hachijo-Island, TOKYO	7.20	1.11	1.54	$3.9 \pm 0.31$	$3.5 \pm 0.28$	63 $\pm 1.0$	$41 \pm 0.6$
Nishikawa-machi, NIIGATA	7.57	1.18	1.69	$1.9 \pm 0.26$	$1.6 \pm 0.22$	$3.3 \pm 0.27$	$2.0 \pm 0.16$
Katsuyama, FUKUI	7.52	1.14	1.64	$2.0 \pm 0.27$	$1.8 \pm 0.24$	$2.5 \pm 0.24$	$1.6 \pm 0.15$
Nose-machi, OSAKA	7.50	1.18	1.60	$0.6 \pm 0.20$	$0.5 \pm 0.17$	$3.8 \pm 0.27$	$2.4 \pm 0.17$
Takamiya-machi, HIROSHI <b>M</b> A	6.83	1.29	1.48	$0.5 \pm 0.19$	$0.4 \pm 0.14$	$0.6 \pm 0.15$	$0.4 \pm 0.10$
Kochi, KOCHI	7.29	1.16	1.55	$2.0 \pm 0.27$	$1.8 \pm 0.23$	$98 \pm 1.2$	$63 \pm 0.8$
Kajiki-machi, KAGOSHIMA	7.58	1.20	1.60	$1.7 \pm 0.26$	$1.4 \pm 0.21$	$26 \pm 0.6$	$16 \pm 0.4$
June, 1986							
Fukuma-machi, FUKUOKA	7.38	1.17	1.68	$1.0 \pm 0.23$	$0.8 \pm 0.19$	$2.7 \pm 0.25$	$1.6 \pm 0.15$

# (3)-3 Strontium-90 and Cesium-137 in Milk(consuming districts) (from Dec. 1985 to Jul. 1986)

Table (3)-3: Strontium-90 and Cesium-137 in Milk

Lasaktan	Component			9 0	Sr	<sup>137</sup> Cs		
Location	Ash(g/l)	Ca(g/l)	K(g/l)	pCi/l	S.U.	pci/l	c.u.	
December, 1985								
Matsue, SHIMANE	7.18	1.11	1.51	$1.8 \pm 0.25$	$1.6 \pm 0.22$	$2.9 \pm 0.24$	$1.9 \pm 0.16$	
Yonagusuku-mura, OKINAWA	7.15	1.10	1.55	$0.4 \pm 0.19$	$0.4 \pm 0.17$	$0.8 \pm 0.17$	$0.5 \pm 0.11$	
January, 1986								
Fukushima, FUKUSHIMA	7.56	1.19	1.66	$1.1 \pm 0.21$	$0.9 \pm 0.17$	$0.9 \pm 0.18$	$0.5 \pm 0.11$	
Osaka, OSAKA	7.48	1.15	1.57	$0.5 \pm 0.19$	$0.4 \pm 0.16$	$0.6 \pm 0.14$	$0.4 \pm 0.09$	
Hiroshima, HIROSHIMA	7.22	1.09	1.56	$1.0 \pm 0.22$	$0.9 \pm 0.20$	$0.8 \pm 0.16$	$0.5 \pm 0.10$	
February, 1986								
Sapporo, HOKKAIDO	7.27	1.19	1.57	$1.9 \pm 0.23$	$1.6 \pm 0.20$	$1.8 \pm 0.21$	$1.1 \pm 0.13$	
Yamagata, YAMAGATA	7.55	1.18	1.66	$1.0 \pm 0.21$	$0.8 \pm 0.18$	$1.1 \pm 0.18$	$0.7 \pm 0.11$	
Shinjuku, TOKYO	6.98	1.11	1.53	$0.7 \pm 0.19$	$0.6 \pm 0.17$	$0.8 \pm 0.16$	$0.5 \pm 0.10$	
Niigata, NIIGATA	7.67	1.19	1.60	$2.0 \pm 0.30$	$1.6 \pm 0.26$	$1.2 \pm 0.22$	$0.8 \pm 0.14$	
Fukui, FUKUI	7.30	1.08	1.60	$2.4 \pm 0.31$	$2.2 \pm 0.29$	$1.5 \pm 0.22$	$0.9 \pm 0.14$	
Nagano, NAGANO	6.82	1.20	1.44	$0.8 \pm 0.27$	$0.7 \pm 0.23$	$1.1 \pm 0.20$	$0.7 \pm 0.14$	
Shizuoka, SHIZUOKA	6.93	1.07	1.48	$0.7 \pm 0.27$	$0.6 \pm 0.25$	$1.7 \pm 0.23$	$1.1 \pm 0.16$	
Nagoya, AICHI	7.38	1.15	1.58	$1.0 \pm 0.37$	$0.9 \pm 0.32$	$0.8 \pm 0.24$	$0.5 \pm 0.15$	
Wakayama, WAKAYAMA	7.99	1.10	1.74	$0.8 \pm 0.22$	$0.8 \pm 0.20$	$0.5 \pm 0.15$	$0.3 \pm 0.09$	
Yonago, TOTTORI	7.39	1.12	1.58	$0.9 \pm 0.21$	$0.8 \pm 0.19$	$1.8 \pm 0.22$	$1.2 \pm 0.14$	
Okayama, OKAYAMA	7.12	1.09	1.53	$0.7 \pm 0.21$	$0.7 \pm 0.19$	$0.6 \pm 0.14$	$0.4 \pm 0.09$	
Yamaguchi, YAMAGUCHI	7.29	1.10	1.59	$1.2 \pm 0.23$	$1.1 \pm 0.21$	$1.3 \pm 0.18$	$0.4 \pm 0.03$	
Matsuyama, EHIME	8.46	1.19	1.61	$1.2 \pm 0.27$	$1.0 \pm 0.22$	$0.0 \pm 0.15$	$0.0 \pm 0.09$	
Kochi, KOCHI	7.14	1.07	1.54	$1.2 \pm 0.26$	$1.1 \pm 0.24$	$0.8 \pm 0.18$	$0.5 \pm 0.12$	
Chikushino, FUKUOKA	7.59	1.16	1.63	$1.2 \pm 0.23$	$1.0 \pm 0.19$	$1.8 \pm 0.22$	$1.1 \pm 0.13$	
Nagasaki, NAGASAKI	7.09	1.08	1.56	$0.6 \pm 0.19$	$0.5 \pm 0.17$	$1.2 \pm 0.19$	$0.7 \pm 0.12$	
Kagoshima, KAGOSHIMA	7.33	1.12	1.56	$1.2 \pm 0.22$	$1.1 \pm 0.20$	$2.2 \pm 0.22$	$1.4 \pm 0.14$	
May, 1986								
Kyoto, KYOTO	7.09	1.06	1.54	$1.0 \pm 0.21$	$0.9 \pm 0.20$	$1.4 \pm 0.18$	$0.9 \pm 0.11$	
Hiroshima, HIROSHIMA	7.10	0.833	1.52	$1.1 \pm 0.22$	$1.3 \pm 0.27$	$1.0 \pm 0.17$	$0.6 \pm 0.11$	
JUly, 1986								
Akita, AKITA	6.61	1.00	1.32	$1.8 \pm 0.24$	$1.7 \pm 0.24$	$23 \pm 0.6$	$17 \pm 0.4$	

### (3)-4 Strontium-90 and Cesium-137 in Milk(powderd milk)

Table (3)-4: Strontium-90 and Cesium-137 in Milk

	Component			•°Sr		<sup>137</sup> Cs	
Location	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	s.U.	pci/Kg	C.U.
June, 1986 *Meiji	7.77	12.4	15.9	34 ± 1.0	2.7 ± 0.08	310 ± 3	19 ± 0.2
JUly, 1986							
Meiji	2.49	3.83	5.18	$2.1 \pm 0.35$	$0.6 \pm 0.09$	$6.6 \pm 0.41$	$1.3 \pm 0.08$
Yukijirushi	2.58	3.53	5.21	$1.9 \pm 0.35$	$0.5 \pm 0.10$	$11 \pm 0.5$	$2.1 \pm 0.10$
Wakodo	2.38	3.57	4.83	$4.2 \pm 0.43$	$1.2 \pm 0.12$	$3.5 \pm 0.31$	$0.7 \pm 0.06$
Morinaga	2.40	3.50	4.80	$2.2 \pm 0.34$	$0.6 \pm 0.10$	$10 \pm 0.5$	$2.1 \pm 0.10$
*Morinaga	8.04	11.6	14.2	$16 \pm 0.7$	$1.4 \pm 0.06$	24 ± 0.8	$1.7 \pm 0.05$

<sup>\*</sup>Skim milk

# (4)-1 Strontium-90 and Cesium-137 in Vegetables(producing districts) (from Nov. 1985 to Feb. 1986)

Table (4)-1: Strontium-90 and Cesium-137 in Vegetables

	Component			90	Sr	137Cs		
Location	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	S.U.	pci/Kg	C.U.	
(Japanese radish)								
November, 1985 Sannohe-machi, AOMORI	0.424	0.244	1.71	5.6 ± 0.29	23 ± 1.2	1.1 ± 0.14	0.7 ± 0.08	
December, 1985 Fukushima, FUKUSHIMA	0.489	0.261	1.92	$3.2 \pm 0.36$	12 ± 1.4	0.4 ± 0.20	0.2 ± 0.10	
January, 1986 Kubokawa-machi, KOCHI	0.450	0.287	1.66	13 ± 0.7	47 ± 2.3	0.3 ± 0.14	0.2 ± 0.09	
February, 1986 Hiroshima, HIROSHIMA Yuya-machi, YAMAGUCHI	0.413 0.653	0.219 0.394	1.62 2.42	$0.7 \pm 0.26$ $4.4 \pm 0.44$	$3.0 \pm 1.2$ $11 \pm 1.1$	0.4 ± 0.16 0.4 ± 0.15	0.3 ± 0.10 0.2 ± 0.06	
(Cabbage)								
November, 1985 Sannohe-machi, AOMORI	0.591	0.425	2.29	12 ± 0.5	27 ± 1.1	2.4 ± 0.21	1.1 ± 0.09	
(Spinach)								
December, 1985 Fukushima, FUKUSHIMA	1.40	0.657	5.35	7.8 ± 0.52	12 ± 0.8	0.2 ± 0.20	0.04 ± 0.04	
January, 1986 Kubokawa-machi, KOCHI	1.72	1.52	5.39	19 ± 0.8	13 ± 0.5	1.5 ± 0.23	$0.3 \pm 0.04$	
February, 1986 Hiroshima, HIROSHIMA Yuya-machi, YAMAGUCHI	1.24 1.88	0.661 0.661	4.66 7.66	$1.5 \pm 0.28$ $4.5 \pm 0.45$	2.3 ± 0.42 6.8 ± 0.69	1.8 ± 0.23 0.7 ± 0.19	0.4 ± 0.05 0.1 ± 0.03	

# (4)-2 Strontium-90 and Cesium-137 in Vegetables(consuming districts) (from Nov. 1985 to Feb. 1986)

Table (4)-2: Strontium-90 and Cesium-137 in Vegetables

Location		Componen	t	90	Sr	137Cs		
	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	S.U.	pci/Kg	C.U.	
(Japanese radish)								
November, 1985 Kanazawa, ISHIKAWA	0.436	0.171	1.73	$3.1 \pm 0.39$	18 ± 2.3	$0.04 \pm 0.14$	0.02 ± 0.08	
December, 1985 Shinjuku, TOKYO Yonagusuku-mura, OKINAWA	0.505 0.646	0.251 0.215	2.05 2.82	$2.6 \pm 0.35$ $3.1 \pm 0.39$	10 ±1.4 14 ±1.8	$0.0 \pm 0.17$ $0.1 \pm 0.16$	$0.0 \pm 0.08$ $0.04 \pm 0.06$	
January, 1986 Nagasaki, NAGASAKI	0.500	0.219	2.15	2.3 ± 0.32	11 ±1.4	0.2 ± 0.14	0.1 ± 0.07	
February, 1986 Yokohama, KANAGAWA	0.500	0.323	1.87	0.5 ± 0.25	1.5 ± 0.77	0.0 ± 0.18	0.0 ± 0.10	
(Cabbage)								
January, 1986 Kumatori-machi, OSAKA	0.562	0.409	2.07	1.5 ± 0.31	$3.7 \pm 0.75$	1.1 ± 0.19	0.5 ± 0.09	
(Spinach)								
November, 1985 Osaka, OSAKA	1.79	0.988	7.29	2.9 ± 0.36	2.9 ± 0.36	1.0 ± 0.20	0.1 ± 0.03	
December, 1985 Shinjuku, TOKYO Kanazawa, ISHIKAWA Yonagusuku-mura, OKINAWA	1.47 1.29 1.64	0.700 0.465 1.10	5.55 5.40 5.73	$2.5 \pm 0.38$ $2.8 \pm 0.35$ $2.5 \pm 0.37$	$3.6 \pm 0.54$ $6.0 \pm 0.76$ $2.2 \pm 0.33$	$\begin{array}{c} 0.4 & \pm 0.24 \\ 1.5 & \pm 0.21 \\ 0.5 & \pm 0.18 \end{array}$	$0.1 \pm 0.04$ $0.3 \pm 0.04$ $0.1 \pm 0.03$	
January, 1986 Nagasaki, NAGASAKI	1.29	0.653	4.43	3.9 ± 0.41	5.9 ± 0.62	0.3 ± 0.18	0.1 ± 0.04	
February, 1986 Yokohama, KANAGAWA	1.57	0.557	6.21	2.5 ± 0.37	4.4 ± 0.66	0.4 ± 0.20	0.1 ± 0.03	

# (5) Strontium-90 and Cesium-137 in Tea(Japanese Tea) (from May 1986 to Jun. 1986)

Table (5): Strontium-90 and Cesium-137 in Tea(Japanese Tea)

Location	Component		•°Sr			137Cs				
	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg		s.u.	р	ci/Kg		c.u.
May. 1986										
Uji, KYOTO	5.01	2.84	19.8	$35 \pm 2.7$	12	±1.0	1100	<b>±</b> 11	54	±0.6
Kaya-machi, KYOTO	4.99	3.25	18.7	$82 \pm 3.9$	25	±1.2	2100	± 15	110	± 1
June, 1986										
Shuzenji-machi, SHIZUOKA	5.02	3.76	17.7	$150 \pm 5$	40	$\pm 1.3$	4100	± 21	230	± 1
Iwata, SHIZUOKA	5.14	3.44	18.9	$45 \pm 3.0$	13	±0.9	950	± 10	50	±0.5
Miyanojou-machi, KAGOSHIMA	6.12	3.72	22.3	$47 \pm 3.4$	13	± 0.9	180	± 5	7.	$9 \pm 0.22$
Chiran-machi, KAGOSHIMA	5.19	3.96	18.2	$38 \pm 2.9$	9.	$.5 \pm 0.74$	290	± 6	16	±0.3

# (6) Strontium-90 and Cesium-137 in Sea Fish (from Oct. 1985 to Jun. 1986)

Table (6): Strontium-90 and Cesium-137 in Sea Fish

	Component			9°Sr		137Cs		
Location	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	S.U.	pci/Kg	C.U.	
(Branchiostegus sp.) November, 1985 Nagasaki, NAGASAKI	1.17	0.508	3.32	0.0±0.32	0.0 ± 0.63	5.5 ± 0.41	1.6 ± 0.12	
(Katsuwonus pelamis) May, 1986 Tosa, KOCHI	1.16	0.100	3.80	0.0 ± 0.22	0.0 ±2.1	19 ± 0.6	5.0 ± 0.15	
(Limanda herzensteini) November, 1985 Mutsu, AOMORI	1.47	1.17	3.52	$0.2 \pm 0.25$	0.2 ± 0.22	3.5 ± 0.30	1.0 ± 0.09	
March, 1986 Ootake, HIROSHIMA	2.14	4.36	3.44	$0.1 \pm 0.23$	$0.03 \pm 0.05$	1.9 ± 0.23	$0.6 \pm 0.07$	
(Scomber japonicus) January, 1986 Sakaiminato, TOTTORI	1.58	1.33	3.56	$0.2 \pm 0.31$	0.1 ± 0.23	$7.1 \pm 0.44$	2.0 ± 0.12	
(Caesio chrysozonus cuvier) December, 1985 Yonagusuku-mura, OKINAWA	3.48	8.45	4.50	$0.4 \pm 0.28$	0.04 ± 0.03	5.5±0.38	1.2 ± 0.08	
(Seriola quinqueradiata) October, 1985 Togi-machi, ISHIKAWA	1.56	1.61	3.69	$0.0 \pm 0.93$	0.0 ± 0.57	5.9±0.79	1.6 ± 0.22	
(Limanda yokohamae) June, 1986 Sendai, MIYAGI	3.87	10.3	3.58	0.5 ± 0.27	0.1 ± 0.03	3.7±0.29	1.0 ± 0.08	
(Sebastes inermis) March, 1986 Ajisu-machi, YAMAGUCHI	3.97	11.3	2.99	0.7 ± 0.26	0.1 ± 0.02	3.5 ± 0.33	1.2 ± 0.11	

Sea Fish

Japanese name	English name	Scientific name			
Amadai	Tilefish	Branchiostegus sp.			
Katsuo	Bonito	Katsuwonus pelamis			
Karei	Flatfish	Limanda herzensteini			
Saba	Common mackerel	Scomber japonicus			
Takasago	Black-tipped fusilier	Caesio chrysozonus cuvier			
Fukuragi	Yellow-tail	Seriola quinqueradiata			
Makogarei Flatfish masbled sole		Limanda yokohamae			
Mebaru	Black Rockfish	Sebastes inermis			

# (7) Strontium-90 and Cesium-137 in Shellfish (from Nov. 1985 to Jun. 1986)

Table (7): Strontium-90 and Cesium-137 in Shellfish

Location	Component			³°Sr		137Cs	
	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	S.U.	pci/Kg	C.U.
(Ostrea gigas) February, 1986 Hiroshima, HIROSHIMA	2.01	0.868	2.42	$0.0 \pm 0.83$	0.0 ± 0.95	0.1 ± 0.47	0.1 ± 0.19
(Turbo cornutus) May, 1986 Ryotsu, NIIGATA	2.31	0.573	2.98	0.0±1.5	0.0 ± 2.6	7.2 ± 0.92	2.4 ± 0.31
June, 1986 Togi-machi, ISHIKAWA	2.75	1.88	2.44	$0.0 \pm 0.71$	0.0 ± 0.38	8.7 ± 0.76	3.6 ± 0.31
(Pecter Yessoensis) November, 1985 Mutsu, AOMORI	2.01	0.321	3.88	$0.0 \pm 0.29$	0.0 ± 0.89	1.9 ± 0.26	0.5 ± 0.07
(Mytilus eduls) June, 1986 Mutsu, AOMORI	2.48	0.795	3.01	$0.0 \pm 0.27$	$0.0 \pm 0.34$	3.9 ± 0.30	1.3 ± 0.10

Shellfish

Japanese name	English name	Scientific name				
Kaki	Oyster	Ostrea gigas				
Sazae Wreath shell Hotategai Scallop		Turbo cornutus Pecter Yessoensis				

# (8) Strontium-90 and Cesium-137 in Seaweeds (from Feb. 1986 to Jun. 1986)

Table (8): Strontium-90 and Cesium-137 in Seaweeds

Location	Component			•°Sr		<sup>137</sup> Cs	
	Ash(%)	Ca(g/Kg)	K(g/Kg)	pCi/Kg	s.u.	pci/Kg	c.u.
(Sargassum fulvellum)							
May, 1986		• • •					
Mutsu, AOMORI	4.23	1.10	14.0	$1.3 \pm 0.32$	$1.2 \pm 0.29$	$4.0 \pm 0.30$	$0.3 \pm 0.02$
(Undaria pinnatifida)							
February, 1986							
Minamichita-machi, AICHI	2.50	0.710	6.58	$1.1 \pm 0.35$	$1.6 \pm 0.49$	$1.0 \pm 0.22$	$0.2 \pm 0.03$
Hiroshima, HIROSHIMA	1.38	0.332	3.88	$0.7 \pm 0.30$	$2.0 \pm 0.91$	$0.8 \pm 0.18$	$0.2 \pm 0.05$
Shimabara, NAGASAKI	2.57	0.681	8.93	$0.7 \pm 0.32$	$1.1 \pm 0.46$	$0.9 \pm 0.21$	$0.1 \pm 0.02$
May, 1986							
Ryotsu, NIIGATA	3.84	1.05	6.41	$0.8 \pm 0.31$	$0.8 \pm 0.29$	$8.5 \pm 0.41$	$1.3 \pm 0.06$
June, 1986							
Sakata, YAMAGATA	2.60	1.05	5.73	$0.9 \pm 0.28$	$0.9 \pm 0.27$	$3.5 \pm 0.26$	$0.6 \pm 0.05$

Seaweeds

Japanese name	English name	Scientific name
Hondawara	Gulfweed	Sargassum fulvellum
Wakame	Wakame seaweed	Undaria pinnatifida



























































